



OCCURRENCE, GEOCHEMISTRY AND INDUSTRIAL QUALITY OF NAFADA GYPSYM DEPOSIT, NORTHEASTERN NIGERIA

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ABSTRACT

Three varieties of gypsum deposits occur in numerous locations within the Senonian Fika Shale in Nafada area, northeastern Nigeria. Geological investigations at Gonja, Warum, Ganko, Shole, Sudingo, Gadi, Mada and Papa mines show the occurrence of the three varieties namely: Satinspar, Selenite and Alabaster. These gypsum deposits occur within the gypsiferous Fika Shale with thicknesses within the span of 1 to 6 cm and depth of 0.2 to 15 meters. The gypsum are hosted within veins sometimes criss - crossing in some locations, pores and fractures inter layered within the fissile Fika Shale intercalated with mud. Analyses of seventeen fresh gypsum samples by X-ray spectrophotometer technique indicate mean average compositions of 93.06, 92.59 and 93.60 wt % of CaSO₄.2H₂O; 2.04, 2.32 and 1.80 wt % SiO₂; 0.38, 0.40, and 0.70 wt % of Fe₂O₃+AL₂O₃; 0.36, 0.34 and 0.11 wt % of MgO; 0.13, 0.14 and 0.13 wt of Alkalis (Na₂O +K₂O) and purity of 96.43, 96.48 and 96.42 for Satinspar, Selenite and Alabaster respectively. These chemical data indicate high grade of gypsum which may be suitable in: cement, ammonium sulphate fertilizer, pharmaceutical, ceramics/pottery, cosmetics, soil amendment, building, chemical, paints and textile industries.

Keywords: Senonian Fika Shale, Gypsum, Satinspar, Selenite and Alabaster.

INTRODUCTION

Nafada is located between latitude 11° 10'N and 10° 05'N and longitude 11° 15'E and 11° 25'E about 118.1 km along Deban Fulani -Bajoga road north of Gombe town in the Gongola Arm. Upper Benue Trough northeastern Nigeria (Fig 1). Gypsum mining sites were assessed around Birin Bolawa, Gadi, Ganko, Gonja, Mada, Papa, Shole, Sudingo, and Warun through minor roads to network of foot paths. Gypsum is a hydrated salt of CaSO4.2H2O in rock form mostly in sedimentary environment associated with shale, limestone, clay and mudstone. Gypsum is an indispensable industrial mineral and raw material in Nigerias' small, medium and major industries for sustainable economic development. The topography of Nafada and environs are generally flat lying and mostly undulating terrain at the mining sites with slope running from the southern to the northern part within an altitude of 259 to 357m above sea level. A sandstone ridge occurs surrounding these terrains at the extreme southern part of the study area otherwise known as the Dumbulwa Bage High described by (Zaborski et al., 1997). Falconer (1911) was the first to carry out mineral survey in northern Nigeria, Carter et al., (1963) first reported the occurrence of gypsum deposits within the sequence of blue black Fika Shale, and was later confirmed by Reyment (1965), who reported that the gypsiferous beds have thickness exceeding 420 meters. Gypsum occurrence in Nafada, Bajoga was reported by Orazulike (1988) and further investigations by Barka (2011) and Tabale (2014) who revealed that the Nafada gypsum forms are Satinspar, Selenite and Alabaster types. These workers confirmed





that the Fika shales are gypsiferous and served as the major host rock for gypsum deposits. Gypsym was also reported by Ntekim (1999) and Mamnan et al (2007) in Guyuk and Cham area within the shale formations in the neighbouring Yola Arm of the Upper Benue Trough. (Fig 2) shows geologic map of the studied area. Gypsum is one of the important raw materials for Nigeria's industrial development especially in cement and chemical industries.

This present work is aimed to assess the industrial quality of Nafada gypsum and environs for suitable application in the various industries like: cement, ammonium sulphate fertilizer, pharmaceutical, ceramics/pottery, cosmetics, soil amendment, building, chemical, paints and textile industries.



Fig I. Map of Nafada showing the study area and Sampling Locations Source: Administrative map of Nafada (2002)









Figure 2: Geologic map of the Upper Benue trough showing the study area.

MATERIALS AND METHODS

Methodology

Chemical compositions of seventeen representative samples of gypsum from different sampling points (Table: 1) within the study area were analyzed at the Ashaka Cement Laboratory, Gombe Nigeria. The samples were brushed, clean and afterwards pulverized in a mechanical crushing machine into a fine powder of 125'microns. To produce pellets used for the analysis, 0.4g of granular stearic acid was mixed with 20g of powdered samples and then re-homogenized in a 'HERZOG' mechanical grinder for about 10 seconds. Then 1g of stearic acid was again added to the reground samples to fill the aluminium pelletizing mould, after which the mould was placed in a 'HERZOG' pelleting machine for 10 seconds. Each of the pellets was analyzed for SiO₂, Fe₂O₃, CaO, MgO, SO₃, K₂O, and Na₂O, combined water and purity (Table 1).



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		Tal	ole 1:	Comp	osition	of stu	died N	afada g	gypsur	n samp	oles (wt %	5)	
Sample Gypsum Chemical parameters (wt %)								/					
	туре	SiO	Al ₂	Fe ₂	CaO	Mg	SO ₃	K ₂ O	Na ₂	Purit	Combin	CaS	CaSO ₄ .2H
		2	O ₃	O ₃		o	5	-	Ō	у	ed	O_4	2 O
										•	water		
G-1	Satins	1.63	0.48	0.14	31.0	0.21	43.2	0.05	0.07	93.0	19.476	74.3	93.78
	par	2	1	0	30	2	79	3	0	50		1	
G-2		1.18	0.30	0.13	30.8	0.29	43.6	0.04	0.07	93.9	19.655	74.5	94.19
		6	3	4	60	9	79	0	0	09		4	
G-3		2.95	0.90	0.28	31.0	0.31	42.1	0.08	0.06	90.6	18.968	73.1	92.13
		7	5	4	09	8	52	7	9	26		6	
G-4		2.39	0.74	0.34	30.5	0.60	42.4	0.07	0.06	91.3	19.123	73.0	92.14
		6	2	0	24	3	96	1	9	65		2	
S-1	Seleni	4.89	1.69	0.81	29.3	0.58	40.8	0.13	0.06	87.9	18.401	70.2	88.66
	te	5	8	0	65	4	91	4	7	16		6	
S-2		1.73	0.50	0.14	30.5	0.21	43.5	0.05	0.06	93.7	19.620	74.1	93.76
		6	8	8	40	8	99	5	9	38		4	
S-3		1.26	0.33	0.13	30.7	0.27	43.7	0.04	0.06	93.9	19.673	74.4	94.16
		6	2	3	54	7	18	5	9	95		7	
S-4		1.55	0.41	0.16	30.6	0.26	43.4	0.05	0.07	93.3	19.537	74.1	93.64
a .		0	1	8	89	9	16	l	0	45	10.005		
S-5		2.13	0.65	0.25	30.5	0.34	42.8	0.06	0.07	92.1	19.295	73.4	92.74
		6	1	8	66	7	17	6	0	85	10 (7 (4	00.06
A-I	Alaba	1.61	0.45	0.25	30.5	0.19	43.6	0.05	0.06	93.9	19.656	74.2	93.86
	ster	8	5	9	23	4	80	1	9	12	10 101	0	01.26
A-2		3.45	1.30	0.53	29.7	0.20	42.4	0.09	0.06	91.2	19.101	/2.1	91.26
		1	5	4	10	Э 0.01	4/	6	9	60 05 5	20.000	0 75 0	05.22
A-3		0.65	0.16	0.09	30./	0.01	44.4 59	0.02	0.07	95.5	20.006	75.2	95.23
A 1		1	3	9	28 20 2	8	58 42.0	8		80	10 201	2	02.44
A-4		2.08	0.03	0.55	30.3 65	0.52	43.0	0.00	0.06	92.5	19.381	/4.0 6	95.44
۸ 5		5 175	5	1	20.2	5	12 8	4	9	99	10 728	0 74 1	02.02
A-3		1.75	0.50	0.11	20.5 20	0.05	43.0 63	0.04	0.07	94.5	19./30	/4.1 0	93.93
16		0.58	0 14	∠ 0.13	29	0.02	03 11 5	0 03	0 07	05.8	20.081	7 75 3	05 42
A-0		0.38	0.14	6	30.7 83	0.02	70	0.03 5	0.07	95.0 15	20.001	6	9 3.4 2
۸.7		0 02	9	0 4 9	30.5	0.01	79 AA 1	0.02	0.07	4J 0/ 8	10.855	0 74.6	94 54
Π-/		0.92 2	3	0.49 4	50.5 65	3	77.1 22	5	0.07	62	19.055	9	94.94
Δ_8		3 29	1 26	- 	29.5	0.07	42 A	0.09	0.06	02 01 3	19116	71 Q	91.10
A-0		8	5	0.07	$\frac{2}{02}$	9	42.4 81	5	8	34	17.110	8	91.10
		0	5	0	02	,	01	Mea	n Value	21		0	
Satinspar		2 04	0.60	0.22	30.8	0.36	42.9	0.06	0.07	96.4	1931	73 7	93.06
		2.04	0.00 7	0.22 9	6	0.50	0	3	0.07	3	17.51	6	25.00
Selenite		2 32	0.72	030	30.3	0 34	42.9	0 07	0.06	96.4	1931	73.2	92.59
		2.32	0	3	8	0.51	0	0	9	8	19.01	8	,2.0,
Alabaster		1.80	0.60	0.38	30.3	0.11	43.5	0.05	0.07	96.4	19.62	73.9	93.60
		1.00	6	2	2		9	4	0	2	17.02	9	22.00
Entire studied		2.05	0.64	0.30	30.5	0.27	43.1	0.06	0.06	96.4	19.41	73.6	93.08
Nafada			4	5	2		3	2	9	4		8	•
Samples													

Sample Identification: G1, G2, G3, G4 = Satinspar Gypsum, S1, S2, S3, S4, S5 = Selenite Gypsum; A1, A2, A3, A4, A5, A6, A7 and A8, = Alabaster Gypsum







Occurrence

Field investigations revealed that the Fika Shales consists of both thick laminated nongypsiferous red clay and gypsum-rich blueblack shale and mudstone and grey coloured highly fissile and friable shale and clay with occasional pebbles where the three varieties (satinspar, selenite and alabaster) of gypsum occurs in pores and fractures within the fissile and friable Fika Shales. Three gypsum varieties are namely: alabaster, satinspar and selenite occur within the Fika Shale at different locations. different depth in Satinspar gypsum forms occur silky, fibrous to acicular crystals as thin beds (averagely 1-5cm) at an average depth of 10.5 meters in Gonja, Warum and Ganko area. It is transparent to translucent, fibrous, silky and acicular crystals commonly aligned parallel to sub-parallel to the bedding plane. Selenite gypsum forms outcrop as thin (averagely \leq 5cm) continuous or discontinuous veins of prismatic crystals arranged vertically or obliquely to the bedding plane. It occurs at an average depth of 9.4 meters at Shole and Sudingo areas. Alabaster gypsum form is massive, fine grained with internal lamination. It is found in Birin Bolawa, Mada and Papa area within the highly friable Fika Shales as thin (averagely ≤ 6 cm) discontinuous veins at an average depth of 7.5 meters.

RESULTS AND DISCUSSION

Geochemistry

During sedimentation processes, certain chemical differentiation involving decomposition of major sulphides (including pyrite and chalcopyrite) and carbonate (including calcite and malachite) minerals and uptake of free radicals leading to process of gypsum formation by dolomitization associated with many other minerals (Murray, 1964; Folk et al., 1974; Milovsky and Kononov, 1984). The various geochemical portioning during sedimentation process tend to fix the released ions from these associated (malachite, smithsonite, anglesite, cerusite) minerals as constituent admixture in the shale rock units. Aided by the high organic matter content of the carbonate rocks and the fine grained nature of the shale units released trace elements (e.g. Cu, Zn, Pb) adhere to the rocks and hence occur as impurities in gypsum bodies (Ntekim and Orazulike 2007).

Industrial Quality

Since water determines the stability of gypsum form, in its natural form, high grade gypsum contains at least 79.1 wt % of CaSO₄ and 20.9 wt % combined water. Therefore, any loss of water by gypsum forms would convert it to anhydrite and thus lower the quality of the mineral. In most cases this loss in water is accompanied by increase in constituent impurities. Relatively (Table 1) the three studied gypsum forms are about the same quality, except that the average combined impurity contents silica, alumina and iron oxide is least in Alabaster type (with 2.79) followed by Satinspar (2.87) and 3.33 for Selenite. Average values of CaSO₄ in the three studied gypsum forms are 73.76, 73.28 and 73.99 wt % for satinspar, selenite and alabaster respectively. This range of CaSO₄ makes the studied gypsum forms slightly below the American Society and Testing of Materials (ASTM) 79.1 wt % CaSO₄ for high grade gypsum (Table 2).

Comparison of Table 1 and Table 2, shows that the average water content in the studied gypsum is 19.31, 19.31 and 19.62 wt % for satinspar, selenite and alabaster respectively, depicting that the studied gypsum may also be regarded as stable because the value slightly fall below the ASTM standard of 20.9 wt % water content for high grade naturally occurring gypsum. According to Holiday (1970) the depth of emplacements of 10.5, 9.4 and 7.5 meters on the average for the studied





Table 2: Average composition of Nafada gypsum relative to standards for various industries 93.60 wt % average CaSO₄.2H₂O, average (MgO) of 0.36, 0.34 and 0.11% and (Na₂O + K₂O) of 0.13, 0.14 and 0.13% [for Satinspar, Selenite and Alabaster forms respectively], can be used for this purpose.

Relative to the British Industrial Specification (BIS) the studied Nafada gypsum meets the BIS requirements for the various industries (Table 2). The BIS specification for cement production is 70-75% CaSO₄.2H₂O [and 80-85% for export quality], 0.6 % (max) K_2O + Na₂O and 3.0% (max) of MgO; the studied gypsum has 93.08 % (max) CaSO₄.2H₂O, 0.13 % K₂O + Na₂O and 0.27 % MgO. BIS specification for ammonium sulphate fertilizer is 85-90 % CaSO₄.2H₂O, 6.0 % SiO₂, 1.0 % Fe₂O₃/Al₂O₃ and 1.5 % MgO, and the studied gypsum has 93.08 % CaSO₄.2H₂O, 2.05 % SiO₂, 0.49 % Fe₂O₃/Al₂O₃ and 0.27 % MgO; meaning that the studied gypsum is therefore suitable for ammonium sulphate fertilizer surgical/pharmaceutical production. For industry the BIS requires 96.0 % CaSO₄.2H₂O, while the studied gypsum has 93.08 % CaSO₄.2H₂O which is slightly below, but could be beneficiated to meet up with the BIS standard. Pottery has BIS required specification of 85 % CaSO₄.2H₂O, 1.0 % Fe₂O₃/Al₂O₃ and 1.5 % MgO, while the studied gypsum has 93.08 % CaSO₄.2H₂O, 0.49 % Fe₂O₃/Al₂O₃ and 0.27 % MgO; hence

is within the specified BIS requirement. Cosmetics manufacture has BIS specification of 75.25 % CaSO₄.2H₂O, 3.0 % (max) SiO₂, and the studied gypsum has 93.08 % CaSO₄.2H₂O, and 2.05 % SiO₂, so can be used for cosmetic manufacture. The studied gypsum with 93.08 % CaSO₄.2H₂O averagely has met the specified limits for materials needed for soil amendment, building, chemical, paints and textile industries (BIS, 1997).

Kurt, (1984) opined that gypsum in its natural form usually contains impurities, probably due to varying amounts of shaly materials, silica, alumina, magnesia, iron oxide and other compounds, thereby lowering the grade of the mineral. The three studied Nafada gypsum forms show average purity of 96.43, 96.48 and 96.42 % CaSO₄.2H₂O for Satinspar, Selenite and Alabaster respectively (Table 2). This rating portends materials suitable for use in the various manufacturing industries hence, Nafada gypsum is regarded as of good quality for industrial applications.

CONCLUSION

Gypsum occurs in various locations within the blue black Fika Shales at Nafada area. Three gypsum varieties (Satinspar, Selenite and Alabaster) were identified and their geochemical assessment reveals that the gypsum were of high grade with an average purity of 93.06, 92.59 and 93.60 % for the satinspar, selenite and alabaster respectively. These results indicate that the Nafada gypsum are suitable for industrial application in ammonium sulphate cement. fertilizer. pharmaceutical, ceramics/pottery, cosmetics, soil amendment, building, chemical, paints and textile industries.



Table 2: American Society for Testing and Materials (ASTM 1981), British Industrial

 Specification (BIS 1997) and Raw Material Research Development Council (RMRDC 2005)

Samp	les/standards	Properties (wt %)									
		CaSO ₄ .2H	Purity	Combined	SiO ₂	Fe ₂ O ₃ /	MgO	$Na_2O +$	CaSO		
		2 O		Water		Al_2O_3		K ₂ O	4		
Satinspar		93.06	96.43	19.31	2.04	0.38	0.36	0.13	73.76		
Selenite		92.59	96.48	19.31	2.32	0.40	0.34	0.14	73.28		
Alabaster		93.60	96.42	19.62	1.80	0.70	0.11	0.13	73.99		
Studied Nafada		93.08	96.44	19.41	2.05	0.49	0.27	0.13	73.68		
gypsum											
ASTM		84-100	-	21.9	-	-	<3.03	< 0.603	79.10		
Specifications											
BIS	Cement	70-75	-	-	-	-	3.0	0.6 (max)	-		
							(max)				
	Soil	50 (min)	-	-	-	-	-	-	-		
	amendment										
	AS Fertilizer	85-90	-	-	6.0 (min)	1.0 (min)	50 (min)	-	-		
		(min)									
	Pharmaceuti	96 (min)	-	-	0.7 (min)	0.1 (min)	50 (min)	-	-		
	cal										
	Pottery	85 (min)	-	-	-	1.0 (min)	50 (min)	-	-		
	Building	75 (min)	-	-	-	-	-	-	-		
	Chemical	94 (min)	-	-	-	-	-	-	-		
	Paints	75 (min)	-	-	-	-	-	-	-		
	Textile	82 (min)	-	-	-	-	-	-	-		

ASTM = American Society for Testing and Materials; BIS = British Industrial Specification; Fertilizer = Ammonium sulphate fertilizer; max = maximum amount; min = amount.

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