



# Structural and Mechanical Properties of Cl- and Br-doped WSe<sub>2</sub> Monolayer for Possible Application in Flexible Photovoltaic Panel: An ab-initio study

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# ABSTRACT

This work investigates the structural and mechanical properties of Chlorine- and Bromine-doped WSe<sub>2</sub> Monolayers. Our result reveals that the introduction of Chlorine impurity into the pristine WSe<sub>2</sub> monolayer is more promising under Se-rich conditions due to the smaller formation energy and may be the most favorable replacement of Se. Further, it is noted that the calculated elastic constants certify the Born stability rule, which is an indication that all the doped materials are mechanically stable, may show considerable resistance to linear compression and behave in a brittle manner. Moreover, our findings suggest a strategy to tailor the mechanical properties of the studied materials for various applications. The study of mechanical properties of 2D materials may further advance the development of ultrathin flexible wearable photovoltaic solar panels and other electromechanical systems.

**Keywords:** Mechanical stability, structural stability, Born criteria, WSe<sub>2</sub> monolayer, Chlorine doped WSe<sub>2</sub>, Bromine doped WSe<sub>2</sub>

# INTRODUCTION

Recently, two-dimensional (2D) transition dichalcogenides metal (TMDCs) have attracted enormous attention of researchers due to their excellent physical properties and potential applications in nanodevices and optoelectronics (Roy & Bermel, 2018). These group of materials are generally represented by the formula  $MX_2$  (where M = transition metal and X = chalcogens Se, S or Te) (Yamusa, Shaari, & Isah, 2022). In TMDCs the inter molecular force i.e. W-W and W-Se are strong covalent bonds, whereas the interlayer force is weak Van der Waals force (Lin, Torsi, Geohegan, Robinson, & Xiao, 2021).

Exposure to high temperature and acute chemical environment during manufacture and domestic uses makes Two-dimensional (2D) Transition metal (TMDCs) liable to imperfections such as cracks and pores, these significantly affect their structural and mechanical properties (Malakar, Thakur. 2022), therefore Nahid, & Islam, understanding structural and mechanical properties of 2D TMDCs is very important for predicting their applications. Among these group of crystals WSe<sub>2</sub> monolayer stands alone by having special properties such as high electron mobility(Rawat, Jena, & De Sarkar, 2018) and very good optical response in the visible band (Ji et al., 2019). These properties make 2D WSe<sub>2</sub> promising, and in fact it has been identified as a potential material for flexible electronic devices (Malakar et al., 2022).

Furthermore, investigations of structural and mechanical stability of a crystal compound is essential before it can be incorporated into an electronic or optoelectronic devices. This is important, so as to avoid quick degradation of



the devices performance (Lian, Lu, Niu, Zeng, & Zhan, 2018).

It is generally believed that doping or alloying improves/tunes the structural and mechanical properties of materials (X. Zhao et al., 2020) and (Singh et al, 2021). At times n-type semiconductor is needed, for example as electron transport layer in perovskite solar cell (Lin et al., 2021; Roy & Bermel, 2018). Halogen can serve as electron donors to WSe<sub>2</sub> monolayers to obtain higher carrier mobility n-type semiconductor (Loh, Zhang, Bosman, & Eda, 2021; Mohammed, Shu'aibu, Abdu, & Aliyu, 2023).

Moreover, further development and applications of doped WSe<sub>2</sub> monolayers are restricted by the absence of systematic investigation on these characteristics, though structural and mechanical stabilities determine the long term stability of crystal compounds.

In this work, we focus on investigation on structural and mechanical properties of pure, Br- and Cl-doped WSe<sub>2</sub> monolayer by using first principle technique.

# **COMPUTATIONAL METHOD**

In this work, DFT calculations were carried out using Quantum ESPRESSO code (Giannozzi et al., 2009). Our Br- and Cldoped WSe<sub>2</sub> monolayers are modeled by replacing the Se atom in 3 x3 x 1 WSe<sub>2</sub> monolayers with Br and Cl atom as shown in Fig. 1. These atoms are chosen because the size of Se atom is almost similar to Br and Cl atoms which reduced the change in the super cell (Radzwan et al., 2020). Moreover, doping on chalcogen site is easier experimentally, than on metal site, which may temper with the structural stability of the crystal compound (Lin et al., 2021). The generalized gradient approximation (GGA) with the format of Pardew-Burke-Ernzerhof (PBE) was used for the exchange correlation functional (Itas, Razali, Tata, Idris, & Khandaker, 2023; Itas, Razali, Tata, Kolo, et al., 2023; Perdew, Burke, & Ernzerhof, 1996; Saadu Itas et al., 2023). In our computation, a plane wave basis set with a cut-off kinetic energy of 320 eV was utilized to expand the wave function. Monkhost-Pack Following the scheme (Monkhorst & Pack, 1976), the Brillouin zone is sampled by a  $12 \times 12 \times 1$  k-point mesh for the PBE calculations. Meanwhile, to avoid any artificial interlayer interaction, a larger vacuum layer of 15 Å was introduced along z direction for the pristine and doped-WSe<sub>2</sub> monolayers (Chhana et al., 2021). All the materials lattice constant and the atomic coordinates were fully optimized before carrying out the mechanical properties Broyden-Fletchercalculations using Goldfarb-Shannon (B.F.G.S.) algorithm, until the Hellmann-Feynman force on each atom was below 0.001 eV/Å (W. Zhao, 2021). Thermo-pw package was employed for the elastic properties calculations.

### **RESULTS AND DISCUSSION**

# **Structural Properties**

Structural properties have a very strong relation with other physical properties such as electronic, optical and magnetic properties of material (Roknuzzaman, Alarco, Wang, & Ostrikov, 2021). In this work, 2 x 2 supercell of WSe<sub>2</sub> monolayer with 12 atoms (4 W and 8 Se) is taken as the undoped structure (Figure 1). By replacing one Se atom in the supercell with Br and later with Cl we perform all the calculations. This gives a dopant concentration of 8.33 %.



Figure 1: Crystal structure of (a) Cl-doped and (b) Br-doped WSe<sub>2</sub> monolayers



Figure 2: Energy volume curves of pure (a) Br-doped (b) and Cl-doped WSe<sub>2</sub> (c) monolayers

In order to get the equilibrium, ground state structural properties of pure, Br- and Cldoped WSe<sub>2</sub> monolayers such as the lattice constant, we run structural optimization calculations on the crystal materials in quantum espresso using GGA PBE by minimizing the total energy with respect to volume of the unit cell. The obtained data is fitted to the Birch Mornaghan equation of state (Equation1) (Katsura & Tange, 2019) the calculated equilibrium lattice constants and bulk of modulus are listed in table 2. Figure 2 shows the ground state energy versus volume plots for the pristine and doped crystal compounds of WSe<sub>2</sub> monolayers. The ground state energies are -53.1078, -220.2267 and -223.3221 Ry for pure, bromine and chlorine doped WSe<sub>2</sub> monolayers respectively. These energies indicate that all the materials are structurally stable (Chhana et al., 2021).





Cl-doped  $WSe_2$  monolayer has the lowest ground state energy (-223.32 Ry) which

suggests it is most likely to be the most stable among the studied materials.

$$E_{tot}(a) = E_o + \frac{9V_o B_o}{16} \left\{ \left[ \left(\frac{a_o}{a}\right)^2 - 1 \right]^3 B'_o + \left[ \left(\frac{a_o}{a}\right)^2 - 1 \right]^2 \left[ 6 + 4 \left(\frac{a_o}{a}\right)^2 \right] \right\}$$
(1)

Where  $a_o$  is equilibrium lattice constant,  $V_o$  is the equilibrium volume per atom,  $B_o$  is the bulk modulus at zero pressure and the pressure derivative of the bulk modulus  $B'_o = \left(\frac{\partial B}{\partial P}\right)_T$ 

It was observed that all doped structures maintained their original honeycomb like structure resembling graphene compound, though with slight lattice distortion which could be associated with closer atomic radius of Br and Cl to Se atom in periodic table. Furthermore, the bond length around the dopant atoms lengthen as observed from table

$$E_f = E_{(doped)} - E_{(pure)} + \mu_{se} - \mu_x$$

Where  $E_{(doped)}$ ,  $E_{(pure)}$  are the total energies per supercell of the doped and pristine WSe<sub>2</sub> monolayers, respectively.  $\mu_{se}$  and  $\mu_x$  stand for the chemical potential of Se atom and Br or Cl dopant respectively.

Formation energy is mostly determined by considering the experimental conditions.

$$\mu_w + 2\mu_{se} = \mu_{wse2}$$

Hence,  $\mu_{se}$  is determined from the energy difference between a Se molecule and one WSe<sub>2</sub> unit that is.

$$\mu_{seo} = \frac{1}{2}(\mu_{wse2} - \mu_w)$$

Chlorine doped WSe<sub>2</sub> has the lowest formation energy, as observed from table 3. This suggests that it is most favorable in terms of structural stability. It has been speculated that halogen (Group vii) doping in TMDCs creates instability. This is true, but in bulk and few layers TMDCs and not in monolayers. The fact is that halogens doping in such materials weaken Van der Waals force, which is interlayer not intralayer force.

Table 1 lists the calculated  $E_f$  of the WSe<sub>2</sub> monolayer under both W-rich and Se-rich

1, resulting a small amount of strain around the dopant atom. This could be the reason for the little distortion in the structure.

Structural stability can be studied using formation energy  $E_f$  expression given by (Ding, Xiao, Cui, Di, & Pan, 2023; Tanko et al., 2021).

(2)

Herein, two limiting fabrication conditions have been considered, the W-rich and Se-rich situations. For the W-rich situation,  $\mu_w$  is assumed to be the energy per W atom in its stable bulk phase. For the Se-rich,  $\mu_{Se}$  and  $\mu_w$ are related through (X. Zhao et al., 2020)

#### (3)

### (4)

conditions. The table reveals that the incorporation of Cl impurities into the pure WSe<sub>2</sub> monolayer is thermodynamically preferred under Se-rich conditions than that of W-rich conditions owing to the smaller formation energy of Cl-doped WSe<sub>2</sub>.

The fact that  $E_f$  of Br-doped WSe<sub>2</sub> case is the smallest, especially for the Se-rich case confirmed it to be the most favorable replacement of Se with Br.





Table	1: Structural	Properties	of Pure and do	ped WSe <sub>2</sub> n	nonolayers with	in GGA-PBE Fu	nctional
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Compound	Bond length (Å)	Bond Angle (°)		a(a.u)	Formation Energy (Ry)	
	dw-D	<b>O</b> W-Se-W	θw-d-w		W-rich	Se-rich
Pure WSe <sub>2</sub>	*2.552	81.33		6.351		
Br-WSe <sub>2</sub>	2.696	80.67	75.72	12.68	67.71	78.23
Cl-WSe <sub>2</sub>	2.588	80.38	79.17	12.66	65.28	72.34

\*W-Se bond length

### **Mechanical Properties**

Analysis of elastic constants  $(C_{ii})$  is essential for understanding the mechanical properties of the materials. These constants can be calculated using the energy of the strained system, assuming an applied macroscopic stress. The energy of the strained crystal material is given by (Chiodo, Binder, & Bastien, 2011).

$$E_{(V,\varepsilon)} = E_{(V_o,0)} + V_o \left[ \sum_i \tau_i \varepsilon_i \overline{z}_i + \frac{1}{2} \sum_{ij} C_{ij} \varepsilon_i \overline{z}_i \varepsilon_j \overline{z}_j \right]$$
(5)  
(5) is the energy of the unstrained Since all the considered materials are two-

dimensional.

equations

Where  $E_{(V_{\alpha},0)}$  is the energy of the unstrained material with equilibrium volume  $V_o$ ,  $\tau_i$  is element in the stress tensor and  $\mathfrak{Z}_i$  is a factor that takes Voit index into consideration

$$Y = \frac{\left(C_{11}^2 - C_{12}^2\right)}{C_{11}}$$
$$v = \frac{C_{12}}{C_{11}}$$

It should be noted from table 4 and figure 3 that the calculated elastic constants are all positive and they certify the Born stability criteria for monolayer i.e.  $C_{11} - C_{12} > 0$ , which indicates that all the doped material in consideration are mechanically stable.

The results also imply that the doped structures considered in this study are less stiff than the pristine system.

Pugh (Feldman, 1976) proposed a rule used for understanding the brittle or ductile nature of a given material, according to him if G/B > 1.75 the material is ductile while G/B <1.75 indicates brittle material. Our results (Table 2) indicate a G/B value of 0.78187, 0.58351 and 0.8268 for pure, Br- and Cldoped WSe<sub>2</sub> monolayers respectively. This

ratio (v) can be calculated from the following (6)

independent elastic constants i.e. C<sub>11</sub> and C<sub>12</sub>.

The in-plane young modulus (Y) and Poisson

consider

only

two

we

(7)

confirmed that the considered materials show relatively low shear deformation resistance and they all behave in a brittle manner.

Table 3 depicts the calculated elastic constants for both pure and doped WSe<sub>2</sub> monolayers, which are matrix elements obtained from computation run based on equation 5. Among the elastic constants,  $C_{11}$ , C<sub>22</sub> and C<sub>33</sub> are essential, they indicate resistance to linear compression in the a-, band c-direction respectively. It can be observed that Br- and Cl-doped WSe<sub>2</sub> monolayers offered almost equal resistance to linear compression along a-, b- and c-axis.

Poisson's ratio defined as the ratio of transverse contraction strain to longitudinal extension strain under a stretching force (Li,





Jia, Zheng, Peng, & Fu, 2020), can be used to determine the stability of a crystal compound. From table 3 all the considered materials have Poisson's ratio greater than 0.25, suggesting all the materials are structurally stable. Zhao et al (X. Zhao et al., 2020) reported a criterion for identifying brittle and ductile material, they suggested that Poisson ratio less than 0.25 indicates brittle nature, while greater than such value means the material is ductile. Based on this, our observations show that our considered materials are brittle in nature. Figure 3 shows that the three elastic constants (C11, C12 and C44) used for calculating the young modulus and Poisson's ratio decrease in value from pure, Br to Cl doped WSe<sub>2</sub> monolayer respectively.

Finally, the mechanical properties (elastic constants, young, bulk moduli, pressure derivative of the bulk modulus and Poisson ratio) display a strong dependent on dopant atoms (Br and Cl).





Figyre 3: A bar chart comparing the elastic constants of the pure and doped WSe<sub>2</sub> monolayer

Table 3: Elastic c	constants (GPa)	and Poisson	ratio for prist	ine and doped	WSe <sub>2</sub> monolay	yers
					-	

Elastic Constant (GPa)	Compounds			
	Pure-WSe <sub>2</sub>	Br-WSe <sub>2</sub>	Cl-WSe <sub>2</sub>	
C11	37.091	31.505	29.206	
C12	7.268	5.443	4.437	
C22	37.091	31.505	29.206	
C33	20.737	17.055	17.172	
C44	11.955	10.479	10.679	
Y	3.5666	2.9828	2.8532	
V	0.1959	0.1727	0.1519	





## CONCLUSION

In this work, we studied the structural and mechanical properties of 2D Br and Cl-doped WSe<sub>2</sub> monolayers. The structural property calculation reveals that the lattice constants calculated are in close agreement with other theoretical and experimental data. The mechanical properties obtained showed that the considered materials are mechanically and exhibit structurally stable. considerable resistance to linear compression and behave in a brittle manner. This suggests that the materials may be promising for photovoltaic and other applications.

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